

## General Chemistry II Final Exam Study Guide

### Exam:

- American Chemical Society
- Friday May 10th @ 8am in SEA 100
- 110 minutes
- 70 multiple choice (A-D)
- Chapter 11-19
  - In order of chapters
- 100 points -- average 35/70 (50%) :')

### Practice:

<http://infohost.nmt.edu/~mheagy/UnOfficialACSPpracticeTest01A.pdf>

### Chapter 11 (2 Questions): Intermolecular Forces and Properties of Liquids

- Dispersion forces (everything)
  - Strength of forces based on sizes
    - I2 bigger than F2 -- I2 has stronger dispersion forces
- Dipole- dipole forces (polar)
- Hydrogen bonding (H bonded to O, F, N)
- Properties of Liquids
  - Boiling point = stronger intermolecular forces
  - Vapor pressure = weaker intermolecular forces
  - Example
    - I2
      - Stronger dispersion forces
      - Higher boiling point
    - F2
      - Weaker dispersion forces
      - Higher vapor pressure

### Chapter 12: Solutions and Colligative Properties

- Concentration units (molarity/molality/percent by mass)
- Colligative properties
  - Boiling point elevation
    - Solution boils at slightly higher temperature than solvent
    - $\Delta T_b = i(\text{Van't Hoff Factor}) \cdot k_b(\text{boiling constant}) \cdot m(\text{molality})$

# General Chemistry Ii Study Guide

**TD Snyder**



## General Chemistry II Study Guide:

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