General Chemistry Study Guide Chapters 1-3

Chapter 1

Key Concepts:

- 1. Chemistry is the study of matter and its properties.
- 2. Matter is anything that has mass and occupies space.
- 3. Mass is the amount of matter in an object.
- 4. Volume is the amount of space an object occupies.
- 5. Density is the mass of an object divided by its volume.
- 6. Elements are the simplest form of matter and are made up of atoms.
- Atoms are the smallest particle of an element that has the chemical properties of that element.
- 8. The periodic table is a list of the elements arranged in order of increasing atomic number.
- 9. Compounds are molecules made up of two or more atoms of different elements.
- Mixtures are combinations of two or more substances that are not chemically bonded together.

Important Events:

- In 1687, Sir Isaac Newton published his law of gravity, which explained the force that attracts objects to one another.
- In 1766, Henry Covendish discovered that water is made up of two elements, hydrogen and oxygen.
- In 1808, John Dalton proposed his atomic theory, which stated that all matter is made up of atoms.

General Chemistry Practical Study Guide Answers

Albert A Gayle

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